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EXAMINER

EPPS FORD, JANET L

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PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

DETAILED ACTION

Continued Examination Under 37 CFR 1.114

1. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office Action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed 4-21-08 has been entered.
2. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office Action.
3. Claims 1, 3-7, and 10-20 are presently pending.

Response to Arguments

4. Applicant's arguments with respect to claims 1, 3-7, and 10-20 have been considered but are moot in view of the new ground(s) of rejection. Applicant's submission of Declarations under 37 CFR 1.132 was sufficient to overcome the rejections set forth in the prior Office Action.

Claim Rejections - 35 USC § 102

5. The rejection of claims 1, 3-7, and 10-13 under 35 U.S.C. 102(e) as being anticipated by Rozema et al. (US Patent NO. 7,019,113), is withdrawn in response of Applicant's providing a showing under 37 CFR 1.132 that any invention disclosed but not claimed in the reference was derived from the inventor of this application and is thus not the invention "by another."

6. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

7. Claims 1, 3-5, 7, 10-15, 17, and 19-20 are rejected under 35 U.S.C. 102(b) as being anticipated by Wolff (WO200075164 A1).

8. WO 200075164 teach the following at page 7:

Polymers for Drug and Nucleic Acid Delivery

Polymers are used for drug delivery for a variety of therapeutic purposes. Polymers have also been used in research for the delivery of nucleic acids (polynucleotides and oligonucleotides) to cells with an eventual goal of providing therapeutic processes. Such processes have been termed gene therapy or anti-sense therapy. One of the several methods of nucleic acid delivery to the cells is the use of DNA-polycation complexes. It has been shown that cationic proteins like histones and protamines or synthetic polymers like polylysine, polyarginine, polyornithine, DEAE dextran, polybrene, and polyethylenimine may be effective intracellular delivery agents while small polycations like spermine are ineffective. The following are some important principles involving the mechanism by which polycations facilitate uptake of DNA:

Polycations provide attachment of DNA to the cell surface. The polymer forms a cross-bridge between the polyanionic nucleic acids and the polyanionic surfaces of the cells. As a result the main mechanism of DNA translocation to the intracellular space might be non-specific adsorptive endocytosis which may be more effective than liquid endocytosis or receptor-mediated endocytosis. Furthermore, polycations are a convenient linker for attaching specific ligands to DNA and as result, DNA- polycation complexes can be targeted to specific cell types.

At page 23 of this reference, compound delivery systems comprising polymers containing pH-labile groups are described as follows:

In some preferred embodiments of the present invention, nucleic acids are delivered to cells by a polymer complex containing a labile group, or groups, that undergoes chemical transformation when exposed to the low pH environment of an endosome. Such complexes provide improved nucleic acid delivery systems, as they provide for efficient delivery and
25 low toxicity.

At page 25 of this reference biologically active compounds containing pH-labile bonds, Described in the following paragraphs on page 26 of the reference:

The invention specifies compounds of the following general structure: A-B-C wherein A is a biologically active compound such as pharmaceuticals, drugs, proteins, peptides, hormones, cytokines, enzymes and nucleic acids such as anti-sense, ribozyme, recombining nucleic acids, and expressed genes; B is a labile linkage that contains a pH-labile bond such as acetals, ketals, enol ethers, enol esters, amides of 2,3-disubstituted maleamic acids, imines, imminiums, enamines, silyl ethers, and silyl enol ethers; and C is a compound. In one embodiment C is a compound that modifies the activity, function, delivery, transport, shelf-life, pharmacokinetics, blood circulation time in vivo, tissue and organ targeting, and sub-cellular targeting of the biologically active compound A. For example, C can be a hydrophilic compound such as polyethylene glycol to increase the water solubility of relatively hydrophobic drugs (e.g. amphotericin B) to improve their formulation and delivery properties. In other embodiments, B is a labile linkage that contains pH-labile bond such as acetals, ketals, enol ethers, enol esters, amides, imines, imminiums, enamines, silyl ethers, and silyl enol ethers.

Page 32 of this reference describes transfection reagents that mediate entry of oligonucleotides/polynucleotides into cells, this reference includes polyamines, and peptides such as membrane active compounds such as melittin described on page 38.

The polymers of this reference also includes amphipathic polymers within the context of their delivery compounds, see page 37.

Pages 51-52 of this reference describes the modification of amine functions in the polymers of the present invention with compounds such as can anhydride (such as

maleic and succinic anhydride) to form an amide acid, this reference teaches that the product of succinic anhydride and a primary amine, reverses back to an amide and anhydride, this reaction is pH sensitive.

The above description of this reference reads on claims 1, 3-5, 7, 10-15, 17, and 19-20.

Claim Rejections - 35 USC § 103

9. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

10. Claims 1, 3-7, and 10-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Wolff as applied above, in view of Mathiowitz et al. (US 6248720) and Haines et al. (US6479464).

11. Wolff does not teach the use of polyvinyl ether in their transfection complexes. Mathiowitz et al. teach that polyvinylethers are functionally equivalent polymers useful for the transfection of nucleic acid into cells.

12. Wolff does not teach the use of paradaxin in their transfection complexes. Haines et al. teaches the use of the fusogenic amphipathic peptide sequence paradaxin as a ligand which serves to promote cellular uptake of nucleic acid by disrupting cellular membranes.

13. It would have been obvious to the ordinary skilled artisan to modify the polymer portion of the transfection compounds of Wolff with the polymers of Mathiowitz et al. and

Haines et al. since these references teach the usefulness of polymers such as polyvinylether (Mathiowitz et al.) and paradaxin (Haines et al.) in the transfection of nucleic acids into cells. Therefore, it would have been obvious to substitute art recognized nucleic acid transfection polymers for the polymers described in Wolff since the prior art polymers are disclosed as functionally equivalent to those polymers described in Wolff.

14. See MPEP § 2144.06 [R-6].II., which describes the obviousness of “[S]UBSTITUTING EQUIVALENTS KNOWN FOR THE SAME PURPOSE:

15. In order to rely on equivalence as a rationale supporting an obviousness rejection, the equivalency must be recognized in the prior art, and cannot be based on applicant’s disclosure or the mere fact that the components at issue are functional or mechanical equivalents. In re Ruff, 256 F.2d 590, 118 USPQ 340 (CCPA 1958) (The mere fact that components are claimed as members of a Markush group cannot be relied upon to establish the equivalency of these components. However, an applicant’s expressed recognition of an art-recognized or obvious equivalent may be used to refute an argument that such equivalency does not exist.); ** Smith v. Hayashi, 209 USPQ 754 (Bd. of Pat. Inter. 1980) (The mere fact that phthalocyanine and selenium function as equivalent photoconductors in the claimed environment was not sufficient to establish that one would have been obvious over the other. However, there was evidence that both phthalocyanine and selenium were known photoconductors in the art of electrophotography. “This, in our view, presents strong evidence of obviousness in substituting one for the other in an electrophotographic environment as a photoconductor.” 209 USPQ at 759.)”

Conclusion

16. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

17. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Janet L. Epps-Smith whose telephone number is 571-272-0757. The examiner can normally be reached on M-F, 10:00 AM through 6:30 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Joseph Woitach can be reached on 571-272-0739. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Janet L. Epps-Smith/
Primary Examiner, Art Unit 1633